

### **Motivation for Project**

- 2,4-Dichlorophenoxyacetic acid (2,4-D) is an herbicide that is used in many products to eliminate weeds on land and in water. This product negatively affects water quality **by** contaminating surface waters, resulting in adverse effects on human and animal life.<sup>1</sup>
- This study is to investigate the effectiveness of removing this herbicide from water using a photocatalytic process.



### What is Photocatalysis?



Photocatalysis is a process which utilizes light energy to cause chemical reactions to speed up. Energy from sunlight is used **an eco-friendly** manner as it uses renewable energy.

### **The Photocatalyst**

- $CuBiW_2O_8$  (CBTO) is a semiconductor that serves as a photocatalyst.
- It is created through a multi-step solid state synthesis.
- It is a favorable photocatalyst as it has a strong energy absorption in the whole visible light range.



### **References**

1. Kwan, C. Y., & Chu, W. (2003). Photodegradation of 2,4-dichlorophenoxyacetic acid in various iron-mediated oxidation systems. Water Research, 37(18), 4405–4412. 2. Yilmaz Akkaya, C., Dombrowski, J. P., Colin-Ulloa, E., Titova, L. V., Lawton, T. J., Alexander, T. E., Brack, E., Drew, C., & Rao, P. M. (2022). Improved synthesis and transient absorption spectroscopy of CuBiW2O8 with demonstration of visible-light-driven photocatalysis and mechanistic insights. Journal of Materials Chemistry A, 10(46), 24888–24895.

# Investigation into the Effectiveness of CuBiW, $O_{g}$ in Photocatalytic Degradation of an Herbicide

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### **The Investigation**



4h @ 800 ° C

Ar flow Cu Box

CuBiW<sub>2</sub>O

- Previous studies have shown that CBTO could degrade a form of the element Chromium ( $Cr^{6+}$ ), a mutagenic and carcinogenic water contaminant, into a 🔰 form much less harmful.<sup>2</sup>
- Using photocatalysts such as TiO<sub>2</sub> and CBTO, a photoreduction study was completed **to investigate if a** similar degradation of 2,4-D would occur.
- The samples were subjected to simulated sunlight for three hours after being stirred in the dark.
- A standard curve of absorbance for the pesticide was created as a reference to compare the testing results.
- Absorption of light was measured using a spectrophotometer and plotted against time.



## **Data Analysis**









• Multiple trials were conducted with different concentrations of the photocatalysts.

 Measurements of this technique revealed the smaller particles of the photocatalysts did begin to break down the pesticide. • The larger particle

photocatalysts did not degrade the pesticide as expected.



The aim of this project is to use clean energy for the purpose of breaking down a pesticide, a water contaminant. Therefore, two of the United Nations Sustainable Goals are targeted.

- respond to the 2,4-D as readily as  $Cr^{6+}$ .
- measuring larger particles of the photocatalyst, as the light appeared to

- understand human impact on Earth.
- service announcement.
- the **Engineering Design Process**.

- support during this investigation.
- Taylor, and Mia Dubosarsky.



# SUSTAINABLE GEALS

### **Final Thoughts**

Results indicate the photocatalyst did not The measurement procedure was not effective for

scatter through the nontransparent samples. A test which more directly measures the herbicide absorbance is recommended. Repeated testing under different conditions such the pH of the water, concentration of

herbicide, size of the particles of the photocatalyst, and concentration of photocatalyst is also recommended.

### In The Classroom

Research a city that is in a climate change crisis to

Inform and encourage the use of renewable energy sources to limit pollution on land and **in water** in own community through a public

Create a water filtration prototype by following

Write a letter to local stakeholders in school and community to fund a green energy initiative.

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AFFORDABLE AND CLEAN ENERGY -0-